# Whole–rock elemental abundances in sandstones and mudrocks from the Tanabe Group, Kii Peninsula, Japan

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#### Abstract

Siliciclastic sediments of the Lower Miocene Tanabe Group crop out in southwest Kii Peninsula, Wakayama Prefecture, and represent a forearc basin succession deposited on the uplifted Shimanto accretionary complex. This report contains whole–rock XRF data for a stratigraphic collection of 87 sandstones and mudrocks from the Tanabe sequence. Fifty–four samples have also been analysed for REE and additional trace element by solution ICP–MS after acid digestion and alkali fusion. The results show characteristic elemental sorting fractionation between sandstones and mudrocks in the members of the basal Asso Formation, which becomes more marked in the overlying Shirahama Formation, probably as a result of winnowing in a storm–dominated shelf depositional environment. Elemental abundances generally compare well with those of the Shimanto protoliths from which the Tanabe Group was derived.

#### Introduction

The Lower Miocene Tanabe Group outcrops extensively in southwest Kii Peninsula, Wakayama Prefecture (Fig. 1). The Tanabe succession has a maximum thickness of about 1500 m, and comprises a varied assemblage of siliciclastic sediments including mudstone, sandstone – mudstone alternations, massive sandstone, conglomerate and breccia (Tanabe Research Group 1984). The Tanabe Group rests unconformably on the Otonashigawa and Muro Groups of the Shimanto terrane, and was deposited in a shallow–water forearc basin, similar to modern equivalents at the SW Japan Pacific margin.

The Tanabe sequence represents the Miocene stage of



Fig. 1. Location map showing distribution of the Tanabe Group in Kii Peninsula and sample sites.

evolution of the inboard and underlying Shimanto accretionary wedge. This report contains whole rock XRF and ICP-MS analyses of a stratigraphic collection from the Tanabe Group, to complement earlier sampling from the Shimanto terrane proper (Roser et al. 1998a). The purpose of both collections is to examine the Cretaceous to Miocene evolution of the SW Japan subduction margin. The main aim of this report is to present the data, briefly outline the analytical techniques used, and describe broad elemental abundances and variations within the Tanabe Group. Analysis of a number of elements by both XRF and ICP-MS also allows comparison between the techniques. More detailed discussion of provenance signatures, effects of recycling, and the influence of depositional environment on the geochemistry of the sequence will be published elsewhere (Roser et al. in prep). Preliminary interpretation has been presented by Roser et al. (2000).

#### **Geological Outline and Sample Suites**

# 1. Tanabe Group

The Tanabe Group has been studied in detail by the Tanabe Research Group (TRG). They mapped the area, refined the stratigraphy, carried out detailed facies analysis, and constructed an integrated basin model, particularly of the lower half of the sequence (TRG 1984, 1985, 1992, 1993). The geological summary below is largely drawn from that work, supplemented by field observation during sample collection.

The Tanabe Group is subdivided into the Asso and Shirahama Formations. The basal Asso Formation is further divided into the Iwayadani, Maro, Hisogawa, and Gohchidani Members, in ascending order (Fig. 2). The Iwayadani Member rests unconformably on Shimanto terrane sediments, and consists of subangular to angular

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Fig. 2. Schematic stratigraphy of the Tanabe Group, after Tanabe Research Group (1992, 1993).

with sandy matrix, sandstones, and conglomerates mudstones, and occasional coaly horizons. The Iwayadani Member thins from north to south, and has been interpreted as drowned valley fill. It is succeeded by conglomerates and mudstones of the Maro and Hisogawa Members, which occur only in the north of the Tanabe outcrop area. The Maro Member consists of subangular conglomerates with sandy matrix, along with mudstones. The Hisogawa Member comprises sandy conglomerates, sandstones and siltstones. All three lower members interfinger with mudstones and siltstones of the lower Gohchidani Member to the south. The Hisogawa Member is overlain by upper Gohchidani mudrocks with occasional sandy alternations. Clasts in the Asso conglomerates are dominated by Shimanto sandstones, thought to be of Otonashigawa Group derivation in the Iwayadani Member, and from the Muro Group in the Maro and Hisogawa Members.

Facies arrangements in the Asso Formation are interpreted as a fan delta complex, with the conglomeratic members representing proximal deposition from a northerly source, passing to more distal muds and silts in the lower Gohchidani Member to the south. Distinct stages of genesis, progradation, aggradation and abandonment of the fan delta complex are recognised from facies analysis (TRG 1993). These stages are roughly equivalent to the contrasting facies in the Iwayadani, Maro, Hisogawa, and upper Gohchidani, respectively. The Asso sequence overall is related to a transgressive phase which in turn consists of three short– term phases : that is , transgressive phase (Hisogawa member), highstand phase (Maro and Hisogawa Members) and then transgressive phase (upper Gohchidani Member).

The Shirahama Formation is divided into five broadly defined fining–upwards members, designated S 1 through S 5 (Fig. 2). In contrast to the Asso Formation, the Shirahama Formation consists primarily of bedded sandstone, sandstone–mudstone alternations, and rarer conglomerates,

sandstones and pebbly mudstones. The S 1 Member marks the first appearance of quartzose schist clasts, in contrast to the Shimanto clasts that dominate in the Asso Formation. Quartz rock fragments and acid volcanics are also significant components in the clast populations, and sandstones are subordinate.

Sandstone packets thin to the south and proportions of mud increase (Fig. 2), reflecting a facies arrangement with respect to a NE to WNW source similar to that of the Asso Formation . Although many sandstones show parallel lamination, hummocky cross-bedding and planar crossbedding, other sedimentary structures are common, including swaley cross-bedding; storm deposits are also recognised (TRG 1985; Shimizu & Hisatomi 1993). These features suggest shallow water deposition, including intertidal environments. Mudstone dikes and sills also occur in some areas (Shimizu 1985).

The Tanabe Research Group (1985) described the Shirahama Formation as shallow marine sediments accumulated on the continental shelf during a regressive phase. The Tanabe Group sequence overall is also not dissimilar to the generally shoaling–upward sequence seen in the filling of ridge–confined forearc basins. Lateral facies variations are also similar to such sequences. The stratigraphy of the Shirahama Formation is currently being revised using detailed sedimentary facies analysis (Hisatomi 1998), similar to that carried out earlier for the Asso Formation. This work has recognised a number of upward coarsening megacycles, in contrast to the lithostratigraphic divisions made previously. Because this revision is still in progress, we have here used the older TRG divisions.

# 2. Sample Suites

A total of 87 Tanabe Group sandstones and mudrocks were collected during fieldwork in October – November 1999. The sampling strategy was intended to collect roughly equal numbers of sandstones and mudrocks from each member of both the Asso and Shirahama Formations, to cover the observed range in lithotype. In practice, this proved to be difficult, with finer size grades underrepresented in the conglomeratic horizons in the Asso Formation. To partially overcome this, laterally equivalent mudstones were collected from the more distal lower Gohchidani Member. The majority of the Asso Formation samples were collected from inland exposures in the north of the Tanabe outcrop area, whereas most Shirahama samples were collected from coastal outcrop in the southwest (Fig. 1).

### Sample preparation

Lithified samples were chipped to <10 mm maximum diameter using a manual rock splitter. Many of the mudstones and siltstones were friable or blocky, and no splitting was required. Chip containing veins or surficial oxidation was discarded after washing in running city water to remove loose surface material. The samples were then soaked in deionised distilled water for 24-36 hours, with several changes of water in that time. The cleaned samples were then dried overnight in an oven at 110°C before crushing. The samples were crushed in a tungsten carbide ring mill, in loads of 70-150 g. Maximum mill times were <60 seconds for the sandstones, and as little as 15 seconds for some unlithified siltstones and mudstones. Such mill times are sufficient to produce powders as fine or finer than by agate mortar systems, with no contamination except for tungsten and cobalt (Roser et al. 1998b). Subsamples of the resulting pulps were then dried in an oven at 110°C for at least 24 hours prior to determination of loss on ignition (LOI).

LOI determinations were made by weighing 5-6 g of dried sample into ceramic crucibles, followed by ignition in an electric furnace at 1000°C for two hours. LOI was then calculated from the net weight loss. The ignited material was then manually disaggregated and recrushed in an agate pestle and mortar, and returned to a 110°C oven for at least 24 hours. This ignited material was used for preparation of both the fusion beads for X-ray fluorescence analysis and the solutions for ICP-MS analysis.

#### **Analytical Methods**

Analyses of major elements and 14 trace element were made using a Rigaku RIX-2000 XRF at Shimane University. All analyses were carried out on glass beads prepared in an automatic bead sampler, using an alkali flux comprising 80% lithium tetraborate and 20% lithium metaborate, with a sample to flux ratio of 1:2. Analytical methods, instrumental conditions and calibration follow those described by Kimura and Yamada (1996). Analyses were monitored by repeat analyses of several GSJ standards, from new beads not included in the calibration. Trace element results were also tested by cross-calibration against new analyses of a suite of 14 Shimanto sandstones and mudrocks from Roser et al. (1998a).

Based on the XRF results, 54 samples were selected for analysis of REE and additional trace elements by solution ICP–MS. Samples were analysed in batches of 20 per day, including one blank and one to three GSJ rock standards. Solutions were prepared using a combined alkali fusion and acid digestion procedure (AFAD), which will be described in detail by Kimura et al. (in prep.). Briefly, the method consists of digestion of 0.1000 g oven–dried sample with 1.0 ml HClO<sub>4</sub> and 1.0 ml 40% HF, in Pt crucibles heated to 200°C on a hot plate. This digest is evaporated to dryness over 3 hr, and 0.5 g anhydrous Na<sub>2</sub>CO<sub>3</sub> alkali flux is then added. The crucibles are then heated for 10 minutes in an electric furnace at >900°C. The fluxed samples are then partially dissolved in the Pt crucibles by addition of 5 ml HNO3 and several drops of HF, heated on a hot plate for 10 minutes at 100°C, followed by addition of 10 mls deionized distilled water and 20 minutes further heating. The resulting solutions are then quantitatively transferred to weighed 100 ml polyethylene bottles, with the residue (if any) retained in the crucibles. A third digestion is then carried out by addition of 10 mls HCl, several drops of HF, and 5 mls deionized water, followed by heating at 150°C for at least 30 minutes to dissolve any remaining residue. This last step is important for high field strength element (HFSE) recovery. In the case of the Tanabe samples little residue was present at the beginning of this step, and any present dissolved well before the 30 minutes heating had elapsed. The resulting solution is then quantitatively added to the polyethylene bottles, and the solution made up to 100 ml (by weight) by addition of deionized water. These stock solutions are stable for several weeks at room temperature, and are subsequently diluted 50,000x for conventional solution ICP-MS analysis.

One 100 ml bottle of procedural blank solution was prepared for each daily batch of solution analyses, along with two 10 ml tubes of solution for each unknown sample or standard rock. One ppb equivalent In (indium) was added to the blank solution and all sample tubes for internal standardization. One ppb equivalent (50 ppm equivalent in original samples) of multi-element mixed standard solution (SPEX Co.) was added to one of the 10 ml tubes of each unknown sample. Elemental concentrations in the unknowns were determined by standard addition method using the blank, sample, and standard addition samples described above. The ICP-MS used was the VG PQ3 at Shimane University, operated in normal nebulization mode. Measurement conditions are: RF power of 1350 W (reflection power = 0), nebulizer gas flow rate 0.7 l/min, intermediate gas flow rate 1.3 l/min, and cooling gas flow rate 13 l/min. Peak jumping mode was used, with three points measured for each mass peak at a dwell time of 10 msec. Three 60 sec acquisitions are applied to the blank, sample and standard addition sample. Isobaric oxide molecular interferences were corrected using correction factors determined from mixed standard solutions. Detail of the analytical procedure will be given in a separate paper (Kimura et al., in prep.).

#### **Results and Discussion**

XRF major and trace element results are given in Table 1, listed on a hydrous basis. The ICP/MS results are given in Table 2. Lithology for each sample was estimated from hand specimens using a hand lens and grain size comparator.

# Whole-rock elemental abundances in sandstones and mudrocks from the Tanabe Group, Kii Peninsula, Japan

 Table 1
 XRF major and trace element analyses of Tanabe Group sandstones, siltstones and mudstones, Kii Peninsula. Major elements wt%, trace elements p.p.m. LOI=loss on ignition. LITH=lithology (CS=coarse grained sst; MS=medium sst; FS=fine sst; VFS=very fine sst; CZ=coarse siltstone; ZST=siltstone; MST=mudstone).

SA#	LITH	SiO₂	TiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub> F	e2O3T	MnO	MgO	CaO	Na <sub>2</sub> O	K₂O	P205	LOI	Total	Ва	Ce	Cr	Ga	Nb	_Ni_	Pb	Rb	Sc	Sr	Th	v	Y	Zr
Asso Formation <i>Iwayadani Member</i> IW1 MS 85.15 0.11 7.25 1.88 0.04 0.45 0.39 1.41 1.18 0.02 2.15 100.02 233 19 11 6 1 7 13 41 2.2 70 4.9 17 10 78																											
IW2 IW3 IW4 IW5 IW6 IW7	MS MS CS FS CS CZ	82.23 83.58 79.86 76.90 77.36 71.51	0.09 0.07 0.21 0.30 0.26 0.44	10.14 9.59 9.61 11.41 10.63 13.88	0.43 0.64 1.71 2.51 1.81 3.36	0.00 0.01 0.03 0.03 0.03 0.03	0.17 0.15 0.43 0.60 0.82 0.79	0.05 0.04 1.27 0.73 1.60 0.71	1.90 1.63 1.63 1.44 1.65 1.46	3.45 3.24 2.05 2.52 2.27 3.18	0.01 0.01 0.05 0.05 0.04 0.05	0.90 0.99 2.71 3.26 3.68 4.30	99.37 99.94 99.55 99.75 100.14 99.71	581 516 343 309 337 357	34 18 28 30 24 41	12 8 18 31 18 38	9 9 8 11 10 14	1 1 3 5 5 8	<1 1 10 12 9 17	20 17 15 16 15 19	91 89 69 86 76 111	1.4 1.4 3.5 5.5 5.7 6.6	89 72 113 115 107 122	5.4 4.2 6.1 7.3 5.6 9.2	12 9 32 51 43 67	12 7 11 15 11	74 77 99 114 101 125
Maro MA1 MA2 MA3 MA4 MA5 MA6 MA7 MA8 MA9 MA10 MA11 MA12	Membel MST FS MST CZ VFS VFS-CZ CZ FS CZ CZ VFS	66.18 75.24 66.94 64.80 68.82 72.13 71.95 68.29 71.95 73.21 77.49 78.07	0.46 0.30 0.54 0.60 0.53 0.42 0.43 0.57 0.41 0.61 0.43 0.36	13.78 11.22 14.76 14.73 14.16 11.96 12.53 15.18 11.94 15.63 13.24 12.47	6.23 2.63 5.03 5.15 4.77 3.90 4.16 4.26 4.09 1.22 0.85 0.90	0.11 0.03 0.11 0.04 0.09 0.12 0.09 0.12 0.09 0.12 0.01 0.01	1.48 1.00 1.36 1.61 1.36 1.32 1.27 1.23 1.33 0.70 0.45 0.42	0.46 1.07 0.45 1.55 0.52 0.84 0.74 0.36 0.94 0.07 0.11 0.11	1.80 1.69 1.72 1.64 1.83 1.76 1.71 1.69 1.74 1.92 1.80 1.94	2.93 2.20 3.13 3.24 2.93 2.38 2.46 3.17 2.35 3.14 2.96 2.81	0.07 0.03 0.05 0.09 0.05 0.06 0.05 0.06 0.05 0.03 0.02 0.02	6.03 4.31 5.78 6.48 5.36 4.78 4.62 4.89 4.92 3.17 2.45 2.35	99.52 99.72 99.85 99.91 100.40 99.66 100.03 99.80 99.84 99.71 99.81 99.45	508 348 410 556 337 312 318 355 316 289 491 443	51 36 48 50 33 34 48 34 52 49 44	37 25 49 56 43 41 51 40 45 29 26	12 10 14 15 16 15 12 16 13 13	8 4 10 11 10 5 5 10 7 10 8 6	15 13 21 22 17 19 18 24 20 9 5	15 14 22 15 13 10 15 12 11 17 21	110 84 122 133 109 86 89 111 86 117 109 104	8.8 6.5 10.5 11.2 9.4 8.8 9.0 9.8 9.2 8.9 5.5 4.9	120 108 102 172 124 119 117 119 121 80 108 109	13.6 10.0 13.2 13.3 8.9 8.5 9.3 11.7 9.5 10.9 9.8 8.4	79 50 97 98 91 79 86 95 75 89 59 50	30 18 27 23 17 18 21 17 22 18 16	186 107 161 150 177 123 128 180 102 153 169 138
HISO HS1 HS2 HS3 HS4 HS5 HS6 HS7 HS8 HS9 HS10	FS FS CZ CZ FS CZ FS FS FS MST MST	80.39 71.18 63.71 69.06 75.26 51.44 77.14 78.09 72.14 68.78	0.23 0.25 0.44 0.51 0.29 0.41 0.29 0.25 0.57 0.70	10.24 9.88 13.17 14.34 11.37 11.44 10.99 10.29 16.14 17.45	2.14 3.74 12.13 5.86 3.51 16.54 2.75 2.41 1.84 1.96	0.03 0.08 0.15 0.07 0.06 0.27 0.04 0.05 0.01 0.01	0.58 1.41 1.05 0.82 0.89 2.53 0.73 0.72 0.69 0.76	0.51 3.38 0.21 1.03 1.16 0.54 0.71 0.08 0.08	1.41 0.99 0.83 1.56 1.19 0.65 1.63 1.60 1.51 1.28	2.15 2.14 3.24 3.36 2.37 2.79 2.37 2.22 3.76 4.35	0.04 0.05 0.06 0.04 0.05 0.05 0.05 0.05 0.04 0.04	2.75 6.70 4.99 3.93 4.36 11.83 3.24 3.36 3.68 3.98	100.48 99.80 99.99 99.77 100.38 99.12 99.77 99.73 100.47 99.41	341 335 368 380 346 367 428 405 367 406	33 35 46 56 33 38 34 33 63 65	20 19 38 41 27 41 24 23 40 51	11 9 16 15 12 11 10 8 19 20	6 6 10 7 10 7 7 12 14	9 11 14 13 8 12 13 8 9 7	12 13 20 18 11 14 16 13 26 21	74 75 110 123 86 98 87 80 142 162	3.9 10.4 18.0 10.6 6.7 27.0 5.2 4.6 7.9 6.9	98 96 68 85 71 60 106 106 105 114	5.5 8.9 9.5 12.3 7.2 12.4 9.1 8.9 12.1 13.4	34 49 124 99 55 145 46 83 87	15 20 32 28 20 40 18 17 25 23	98 108 185 182 131 131 112 107 180 182
Goha GH1 GH2 GH3 GH4 GH5 GH6 GH7 GH8 GH7 GH8 GH9 GH10 GH11 GH12 GH13	hidani N VFS CZ VFS CZ MST MST FS VFS CZ CZ CZ CZ CZ CZ	<b>lember</b> 76.16 74.49 75.31 74.23 64.44 60.95 73.93 75.88 69.75 76.79 69.64 68.91 71.18 71.09	0.31 0.43 0.36 0.50 0.78 0.64 0.30 0.40 0.57 0.35 0.55 0.57 0.60 0.67	11.36 13.38 12.07 13.56 16.87 14.38 10.59 12.43 14.62 11.03 13.03 12.92 14.13 14.90	1.87 2.22 3.20 1.97 4.60 7.79 2.34 2.97 5.39 2.92 4.04 4.19 4.54 3.53	0.03 0.02 0.04 0.05 0.08 0.05 0.01 0.01 0.02 0.07 0.07 0.03 0.02	1.01 0.72 0.83 0.71 1.68 2.47 1.11 0.84 1.42 0.75 1.51 1.51 0.95 0.90	1.42 0.44 0.54 0.31 0.85 2.55 0.15 0.12 0.89 1.37 1.40 0.39 0.27	1.73 1.93 1.97 2.03 1.37 1.35 2.09 2.34 1.89 2.29 2.18 2.14 2.04 2.09	2.12 2.75 2.33 4.30 3.54 2.42 2.41 2.68 2.32 2.63 2.61 2.89 3.26	0.07 0.06 0.06 0.07 0.09 0.06 0.07 0.05 0.11 0.08 0.08 0.14 0.09	4.05 3.14 3.36 3.20 5.68 4.56 2.13 3.44 2.24 5.10 4.98 2.92 2.95	100.13 99.59 99.98 99.64 100.14 100.12 100.01 99.62 99.97 99.70 100.19 99.39 99.82 99.78	847 428 399 483 604 549 547 492 604 460 464 534 533	35 47 39 51 69 52 31 46 51 42 51 52 55	28 35 32 45 117 76 32 41 52 36 67 53 57	12 12 13 20 15 10 11 15 11 14 13 14	67681526910 10810111011	13 13 15 60 44 16 14 21 25 26 17 16	17 15 16 20 19 19 13 15 20 23	80 103 88 110 167 138 83 87 103 79 98 98 116 129	5.8 7.4 6.9 7.0 13.9 11.2 6.8 5.1 7.6 5.6 7.6 8.4 9.3 9.5	191 146 150 176 75 89 180 119 93 140 214 214 113 103	7.3 10.9 8.8 10.6 14.2 12.4 6.8 9.7 11.3 6.9 7.9 9.0 12.0 11.1	43 66 62 74 126 117 41 58 98 46 80 82 99 111	15 19 17 28 17 15 25 18 22 23 28 25	160 171 143 196 148 137 142 170 166 216 172 172 181 179
Shir	ahama Jember	Forma	ation																								
S1-1 S1-2 S1-3 S1-4 S1-5 S1-6 S1-7 S1-8 S1-9 S1-10	VFS FS FS MST MST VFS MST VFS MST	82.68 82.61 78.11 81.57 72.76 71.33 78.54 65.75 79.87 65.06	0.19 0.18 0.20 0.21 0.51 0.49 0.25 0.75 0.24 0.57	10.33 10.25 10.02 10.67 14.56 13.77 10.89 17.76 10.54 14.66	0.67 0.51 2.16 1.05 2.38 3.37 1.79 3.69 1.54 5.77	0.01 0.03 0.01 0.02 0.03 0.02 0.02 0.02 0.02 0.04	0.29 0.26 0.82 0.31 0.70 0.95 0.54 1.04 0.57 1.55	0.16 0.20 1.30 0.13 0.20 0.50 0.96 0.21 0.43 0.71	1.72 1.81 1.69 1.80 2.17 2.15 2.44 1.39 2.34 1.80	2.62 2.64 2.49 2.67 3.17 3.38 2.50 4.59 2.34 3.61	0.03 0.02 0.03 0.05 0.05 0.03 0.05 0.03 0.05	1.37 1.30 3.12 1.48 3.42 3.85 2.52 4.61 2.17 5.98	100.07 99.79 99.99 99.93 99.93 99.87 100.48 99.85 100.09 99.81	443 397 402 405 471 528 439 439 432 488	36 39 36 56 54 44 71 44 53	26 23 27 24 50 46 30 72 33 51	9 10 10 14 14 10 21 8 16	4 5 4 13 11 5 16 7 12	6 4 10 8 17 15 10 21 10 18	17 17 17 25 21 19 24 17 19	86 85 81 121 125 84 184 82 135	2.4 1.7 5.4 2.7 6.2 6.4 3.3 11.1 2.9 9.4	93 90 123 91 125 115 138 93 106 120	7.8 6.0 7.3 6.8 12.6 10.3 7.7 16.1 9.5 11.7	19 17 27 21 64 70 30 115 29 85	13 14 19 21 24 16 30 16 25	224 233 272 300 235 206 222 159 262 167
S2-1 S2-2 S2-3 S2-4 S2-5 S2-6 S2-7 S2-8	lember CZ FS VFS VFS VFS VFS MST FS	70.85 74.28 80.61 81.26 81.23 79.70 77.77 80.82	0.52 0.27 0.24 0.22 0.24 0.30 0.42 0.27	14.18 10.74 11.46 10.91 11.08 11.58 12.93 10.44	3.59 2.01 0.67 0.72 0.62 0.76 0.75 0.95	0.03 0.03 0.01 0.01 0.01 0.01 0.00 0.01	0.98 1.08 0.45 0.34 0.33 0.34 0.37 0.33	0.37 2.14 0.46 0.15 0.44 0.22 0.14 0.52	1.56 1.76 2.25 2.23 2.22 2.42 2.24 2.24 2.44	3.54 2.84 2.70 2.64 2.49 2.78 3.14 2.49	0.05 0.04 0.03 0.03 0.04 0.03 0.04 0.03	4.17 4.13 1.68 1.32 1.53 1.46 1.76 1.54	99.84 99.33 100.56 99.84 100.21 99.60 99.56 99.83	455 454 454 417 442 495 541 500	57 41 43 40 42 40 57 43	49 28 29 24 22 24 34 20	14 10 10 10 10 12 9	11 5 7 4 5 8 8 6	15 11 7 4 5 7 4 7	20 19 18 16 19 15 17 14	129 95 89 87 84 95 111 85	7.6 5.7 1.6 1.5 2.3 2.9 3.5 3.4	114 209 108 98 117 113 107 122	12.2 7.8 8.6 7.5 8.3 7.5 10.5 7.6	79 35 24 23 25 33 46 29	25 19 15 16 15 16 19 17	213 217 259 231 304 206 285 271
<b>S-3 M</b> S3-1 S3-2 S3-3 S3-4 S3-5 S3-6 S3-7 S3-8 S3-9 S3-10 S3-MD	lember VFS FS FS VFS MST MST MST MST M SILL ZST M DIKE	80.56 79.56 79.62 80.75 71.28 61.25 70.13 70.34 70.06 67.45 69.15	0.36 0.21 0.19 0.22 0.30 0.65 0.67 0.66 0.32 0.72 0.62	11.54 9.79 10.00 9.18 10.12 15.36 16.23 16.34 10.16 16.84 14.59	0.53 2.21 1.38 1.18 2.74 8.44 3.52 3.14 3.69 4.40 4.81	0.01 0.02 0.02 0.06 0.11 0.03 0.03 0.06 0.03 0.04	0.36 0.60 0.59 0.55 1.62 4.43 1.16 1.09 1.91 1.38 1.64	0.10 0.79 0.97 1.15 4.61 0.86 0.09 0.10 4.57 0.18 1.25	1.79 1.44 1.43 1.39 1.46 1.17 1.44 1.49 1.33 1.27 1.19	3.32 2.66 2.87 2.59 2.41 2.30 3.48 3.64 2.24 3.86 3.13	0.05 0.02 0.02 0.02 0.14 0.04 0.04 0.03 0.05 0.11	1.32 2.65 2.57 2.54 5.30 5.02 3.18 3.17 5.51 3.69 3.92	99.91 99.95 99.64 99.58 99.73 99.97 100.05 99.88 99.86 100.46	590 620 691 577 493 255 430 433 404 434 431	52 29 25 39 63 68 68 68 46 75 59	20 18 13 22 21 56 57 23 61 64	11 9 8 10 17 17 17 10 18 16	8 6 7 13 13 13 5 14 12	2 6 4 7 46 14 19 10 21 26	24 16 18 14 13 21 15 17 14 24 23	117 83 88 82 81 97 143 148 80 158 126	3.1 2.8 3.0 2.5 6.9 11.2 8.8 7.9 8.4 10.8 8.7	110 186 191 183 219 67 60 64 240 61 125	8.9 5.3 7.4 5.0 7.1 12.3 13.5 14.5 7.6 14.7 10.8	36 23 22 20 27 110 86 91 39 105 86	15 11 10 13 22 39 24 25 19 29 26	337 178 127 161 265 202 231 229 242 209 190
<b>S-4 M</b> S4-1 S4-2 S4-3 S4-4 S4-5 S4-6 S4-7 S4-8 S4-9	ember CS MS ZST MS MS ZST FS ZST	63.70 64.72 72.01 66.33 83.63 83.62 64.86 84.44 72.19	0.05 0.09 0.10 0.89 0.14 0.17 0.83 0.12 0.78	5.60 6.52 7.05 17.38 9.09 9.12 16.76 8.27 14.95	0.68 0.83 1.26 1.09 0.19 0.48 1.36 0.81 0.28	0.03 0.02 0.09 0.00 0.00 0.00 0.01 0.00 0.00	4.96 5.55 2.95 0.20 0.14 0.08 0.09 0.07 0.14	10.59 7.36 4.73 0.08 0.02 0.02 0.15 0.01 0.05	0.43 0.66 0.47 1.76 1.66 1.69 1.99 1.52 3.34	1.53 1.82 1.99 10.51 3.91 4.07 11.06 3.86 6.39	0.03 0.09 0.03 0.01 0.00 0.00 0.01 0.00 0.01	11.84 11.95 8.86 1.76 0.78 1.08 2.13 1.28 1.65	99.44 99.62 99.54 100.01 99.57 100.34 99.26 100.39 99.80	412 402 596 1989 811 720 1965 675 1285	17 16 17 35 14 27 41 21 37	3 6 11 40 8 9 29 9 31	5 6 15 9 8 7 14	1 2 19 2 3 18 2 15	3 2 3 <1 <1 9 1	10 15 13 16 19 23 20 18 13	42 55 63 309 120 128 332 125 198	8.4 7.1 5.6 9.8 1.0 1.0 9.3 1.3 8.2	312 155 176 105 79 76 147 70 128	3.7 3.7 4.6 17.3 4.9 5.0 19.2 2.8 12.6	5 6 10 70 14 12 58 12 52	7 9 13 24 8 11 38 6 21	58 80 88 248 125 172 221 112 260
<b>S-5 M</b> S5-1 S5-2 S5-3 S5-4 S5-5 S5-6	ember MST MS-FS MS MS MS mst mst	69.47 84.94 84.87 84.41 66.40 66.23	0.72 0.13 0.15 0.16 0.85 0.88	17.53 8.54 8.96 9.28 18.76 18.38	1.77 0:40 0.51 0.33 2.66 2.39	0.01 0.00 0.01 0.01 0.02 0.02	0.68 0.17 0.18 0.20 0.75 0.84	0.00 0.00 0.00 0.00 0.00 0.00	0.25 0.10 0.05 0.32 0.04 0.06	5.83 4.05 4.37 4.12 6.31 6.02	0.01 0.00 0.00 0.00 0.02 0.02	3.53 1.16 1.06 1.27 4.21 4.18	99.79 99.51 100.17 100.09 100.01 99.01	625 674 726 670 646 551	58 15 23 23 76 88	57 14 18 12 64 67	23 7 8 8 27 24	17 2 4 16 15	4 <1 1 8 9	10 11 12 11 15 10	252 140 150 146 278 282	9.9 0.6 0.5 1.7 12.4 13.5	57 59 63 60 61 43	15.2 5.1 3.4 4.8 15.5 15.5	96 13 12 14 107 124	26 6 9 8 23 25	305 101 141 136 287 244

Table 2 ICP-MS REE and trace element analyses of Tanabe Group sedimentary rocks and GSJ standards included in the same runs. Sample numbers andlithologies as in Table 1. All data are p.p.m., reported as analysed (ignited basis).

SA#	LITH MS	9.28	Ce 19.73	Pr 2 12	7.73	Sm 1.50	Eu 0.41	Gd	Tb 0.22	Dy	Ho 0.25	13 66.0		<u>Yb</u>	Lu	<u> </u>	Ta 0.25	175	U 0.82	<u>Zr</u>	Nb	Be	Sn 0.91	<u>Sb</u>
iW2	MS	22.32	36.04	4.46	15.54	2.55	0.69	2.29	0.28	1.71	0.31	0.81	0.12	0.81	0.12	1.74	0.25	4.87	0.92	59	2.00	0.75	0.50	0.52
IW4 IW5	FS	14.01 16.65	30.44 36.31	3.21	11.78 14.33	2.16	0.50	1.94	0.28	1.66 1.93	0.32	0.90	0.16	1.04 1.38	0.16	2.96 3.42	0.44	5.68 7.01	1.18 1.49	117 149	4.46 5.78	1.04 1.63	1.14 1.45	0.28 0.21
IW7	CZ	20.56	46.11	4.60	16.67	2.92	0.61	2.55	0.36	2.33	0.48	1.39	0.24	1.58	0.25	3.45	0.63	8.92	1.88	124	7.53	1.77	1.86	0.34
MA1 MA2	MST FS	26.68 19.09	59.53 42.07	6.44 4.47	25.18 16.89	5.42 3.31	1.13 0.74	5.41 3.27	0.81 0.49	4.87 2.94	0.96 0.57	2.61 1.58	0.41 0.25	2.78 1.71	0.42 0.26	5.57 3.40	0.88	12.25	2.74 1.85	207 122	10.62	2.34	2.37	0.42
MA5	CZ	23.52	51.95	5.47	20.42	3.95	0.79	3.72	0.54	3.47	0.70	1.99	0.33	2.24	0.36	5.14	0.76	10.30	2.35	190	9.18	1.99	2.04	0.44
MAG MA11	CZ	22.03	46.93	4.18	17.03	2.98	0.73	2.94	0.46	2.83	0.56	1.57	0.25	1.71	0.26	3.44 4.51	0.54	7.69 8.71	1.64 1.91	121 161	6.62 7.89	1.47 1.44	1.69 1.91	0.17 0.58
MA12	VFS	19.76	42.91	4.54	16.52	2. <b>9</b> 8	0.57	2.37	0.34	2.30	0.50	1.49	0.25	1.75	0.27	4.01	0.60	8.51	1.69	154	7.28	1.29	1.86	0.25
HS1 HS4	FS CZ	16.61 26.61	35.64 59.12	3.76 6.34	14.07 23.60	2.69 4.63	0.57 0.90	2.54 4.40	0.37 0.69	2.34 4.42	0.45 0.90	1.28 2.52	0.20 0.40	1.40 2.73	0.21	3.31 5.25	0.50	7.20	1.47	121 192	5.63 10.58	1.33	1.65 2.52	0.18
HS5	FS	18.01	38.23	4.12	15.39	2.99	0.63	2.77	0.43	2.74	0.56	1.54	0.25	1.71	0.27	3.94	0.66	8.10	1.86	141	6.62	1.61	1.84	0.22
HS7 HS9	MST	29.21	39.23 60.39	4.29 6.59	16.20 24.12	3.15 4.09	0.68	3.21	0.44	3.29	0.53	1.44 2.19	0.23	1.56 2.53	0.23	3.24 5.15	0.57 0.91	7.96 12.60	1.60 2.66	116 177	6.48 11.19	1.52 1.93	1.69 2.65	0.24 0.53
GH2	cz	22.12	48.36	4.93	18.02	3.40	0.71	3.00	0.44	2.78	0.57	1.71	0.28	1.97	0.30	5.13	0.66	9.02	2.00	194	8.24	1.69	1.97	0.22
GH4 GH6	CZ MST	24.79 30.68	52.70 67.20	5.83 7.14	21.26 27.22	3.92 5.32	0.79 1.09	3.52 5.02	0.50 0.77	2.98 4.66	0.61 0.93	1.67 2.60	0.27 0.41	1.90 2.75	0.29 0.41	5.21 4.42	0.71	10.10	2.25	205 153	9.06 12.20	1.63	1.98	0.40
GH7	FS	16.96	35.00	3.93	14.45	2.88	0.97	2.82	0.42	2.45	0.48	1.32	0.21	1.44	0.22	3.90	0.50	6.77	1.57	144	6.19	1.08	1.50	0.38
GH8A	CZ	27.31	39.50 57.19	4.37 6.18	22.86	4.09	0.87	3.47	0.55	3.74	0.43	2.27	0.22	2.47	0.25	4.65 4.68	0.61	8.02 10.91	1.84 2.52	175 168	7.60	1.15 1. <b>6</b> 8	1.62 2.24	0.37 0.52
GH12 GH13	CZ CZ	27.47 27.34	59.78 51.50	6.45 5.93	24.79 21.82	4.98 3.95	1.11 0.80	4.71 3.44	0.68 0.53	4.14 3.39	0.81 0.71	2.21 2.04	0.35	2.35 2.31	0.36	4.81 4.59	0.84	10.72	2.38	194 169	10.58	1.84 2.11	2.28	0.50
S1-1	VFS	15.33	31.22	3.39	12.37	2.26	0.55	2.08	0.30	1.99	0.42	1.25	0.21	1.49	0.25	6.29	0.38	6.53	1.77	271	4.68	1.12	4.57	0.33
S1-2	VFS	15.44	32.34	3.47	12.75	2.31	0.51	2.04	0.30	1.93	0.40	1.18	0.21	1.49	0.24	6.63	0.38	6.71	1.76	266	4.58	1.17	1.26	0.23
S1-6	MST	26.20	54.23	5.96	22.27	4.16	0.86	3.85	0.56	3.58	0.71	2.05	0.33	2.34	0.36	5.97	0.80	10.91	2.65	287	10.37	1.88	2.28	0.49
S1-8 S1-9	MST VFS	36.62 17.64	72.91 37.28	8.31 4.06	31.05 15.07	5.60 2.79	0.96 0.63	4.70 2.48	0.72 0.38	4.54 2.39	0.92 0.50	2.62 1.43	0.42 0.24	2.88 1.72	0.44 0.27	4.57 6.63	1.10 0.49	16.16 7.38	3.04 1.93	150 237	14.20 5.67	3.25 1.11	3.29 1.30	0.35
S2-1	CZ	27.77	60.63	6.46	24.08	4.64	0.91	4.08	0.60	3.79	0.76	2.10	0.34	2.37	0.37	5.84	0.90	11.74	2.73	244	10.87	2.26	2.43	0.36
S2-2	FS	18.50	41.15	4.42	16.61	3.35	0.79	3.33	0.52	3.14	0.60	1.63	0.26	1.76	0.28	6.36	0.51	7.93	1.92	265	6.05	1.16	1.38	0.38
S2-6	VFS	18.13	38.49	4.11	14.83	2.80	0.62	2.52	0.38	2.37	0.48	1.40	0.23	1.66	0.25	5.28	0.56	7.75	1.94	203	6.83	1.25	1.46	0.35
S2-7	MST	23.11	49.10	5.28	19.61	3.51	0.71	3.02	0.43	2.75	0.56	1.72	0.29	2.00	0.31	7.53	0.71	9.49	2.26	295	8.70	1.27	1.69	0.37
53-2 53-4	FS	14.37	29.19 34.20	3.79	13.66	2.11	0.56	2.23	0.28	2.03	0.33	1.06	0.15	1.06	0.17	4.59 3.41	0.40	5.09 5.53	1.14	171	4.79 4.71	1.07	1.04 0.99	0.20
S3-5 S3-7	VFS	20.74 34.96	42.13 74.12	4.65 7.79	17.13 29.16	3.35 4.87	1.00 0.86	3.49 4.13	0.51 0.55	3.25 3.48	0.65 0.71	1.83 2.13	0.30	2.04	0.33	7.45	0.56	8.09	1.92	315	6.81 13.92	1.15	1.39	0.39
S3-8	MST	35.00	74.53	7.97	29.80	5.48	0.96	4.84	0.66	4.12	0.82	2.36	0.39	2.72	0.41	6.31	1.05	13.93	2.84	242	13.39	2.25	2.65	0.33
53-9 53-10	ZST	19.57 36.83	40.93 80.67	4.40 8.57	31.68	3.30 6.02	1.11	3.17 5.52	0.49	2.93 4.74	0.58	1.64	0.26	1.82	0.28	5.92 6.18	0.56	7.76 14.85	2.09 3.06	230 236	6.79 14.38	1.37 2.48	1.30 3.08	0.66 0.39
S3-MD	M DIKE	31.30	65.30	7.24	26.56	5.09	1.07	4.75	0.68	4.12	0.79	2.19	0.36	2.43	0.37	5.64	0.95	12.24	2.61	218	11.81	2.20	2.55	0.59
54-3 S4-4	MS	11.50 23.07	23.32 34.48	2.58 4.91	9.45 19.36	1.86 3.43	0.63	3.04	0.30	1.90 3.02	0.38	1.10 2.14	0.18	1.20	0.18 0.47	2.67 6.80	0.31	4.72 18.32	0.90 3.10	102 224	2.98 18.00	1.10 1. <b>3</b> 7	0.68 3.69	0.27 2.25
S4-6	MS 7ST	12.70	25.91	2.88	10.33	1.94	0.51	1.77	0.24	1.39	0.28	0.84	0.14	1.06	0.17	3.77	0.48	5.27	1.13	146	4.20	0.66	0.81	0.57
S4-8	FS	8.02	15.99	1.78	6.32	1.13	0.35	1.03	0.13	0.81	0.16	0.49	0.09	0.68	0.11	2.67	0.37	3.57	0.77	99	3.02	0.56	0.80	0.67
S4-9	ZST	21.94	44.58	4.91	17.47	3.20	0.75	3.01	0.44	2.96	0.66	2.16	0.39	2.94	0.47	7.22	1.44	13.98	3.76	260	16.79	1.43	4.85	2.89
S5-2	MS-FS	9.32	19.08	2.00	7.30	1.35	0.36	1.20	0.16	1.04	0.20	0.63	0.45	0.77	0.12	3.08	0.36	4.03	0.96	118	3.45	2.58	1.01	2.22 0.85
S5-3 S5-4	MS MS	10.15 9.78	18.70 20.05	2.04 2.24	7.31 8.24	1.28 1.51	0.35 0.39	1.18 1.39	0.17 0.20	1.12	0.23 0.25	0.69 0.69	0.11 0.12	0.85 0.88	0.14 0.14	5.20 3.60	0.38 0.42	3.57 4.34	1.00 1.04	207 139	3.80 4.02	0.78 0.75	1.44 1.22	1.20 0.69
S5-5	MST	37.64	85. <b>99</b>	9.18	33.75	6.10	1.11	5.26	0.71	4.20	0.83	2.47	0.42	3.14	0.51	8.02	1.47	18.27	3.87	291	17.36	2.85	5.80	4.48
GSJ STANDARD ROCKS																								
STD JB-1b	Basalt	La 39.75	Ce 68.41	7.16	Nd 26.28	<u>Sm</u> 4.93	Eu 1.49	Gd 5.03	<u>Tb</u> 0.70	<u>Dy</u> 4.15	Ho 0.78	Er 2.06	<u>Tm</u> 0.31	<u>Yb</u> 1.99	0.30	<u> </u>	<u>Ta</u> 1.59	9.35	U 1,57	<u>Zr</u> 123	Nb 24,74	Be 1,27		<u>Sb</u> 0,20
JB-1b		41.51	72.36	7.44	27.71	5.09	1.55	5.20	0.73	4.32	0.80	2.10	0.32	2.10	0.30	3.40	1.63	9.73	1.65	143	26.53	1.22	1.76	0.10
JB-1b		41.43	71.59	7.40	27.44	5.10	1.55	5.00	0.71	4.15	0.81	2.12	0.32	2.00	0.31	3.41	1.68	9.62	1.62	127	25.64 25.53	1.28	1.71	0.15
JB-1b JB-1b		39.08 39.80	68.55 68.25	7.09 7.15	25.75 26.21	4.78 4.95	1.47 1.49	4.81 5.08	0.68 0.69	4.12 4.12	0.77 0.78	2.05 2.05	0.30 0.31	1.98 1.97	0.28 0.30	3.27 3.29	1.57 1.70	9.27 9.39	1.61 1.61	128 133	25.81 25.99	1.32 1.29	1.69 1.79	0.10
Average	-	40.48	70.49	7.29	26.77	5.01	1.51	5.04	0.70	4.19	0.79	2.08	0.31	2.02	0.30	3.35	1.63	9.53	1.62	131	25.71	1.27	1.70	0.14
StaDev CV%		2.12 5.24	4.78 6.78	4.76	1.58 5.91	6.18	4.01	0.27 5.30	4.48	3.97	4.56	2.78	4.37	5.13	7.04	0.12 3.60	0.10 6.14	0.45 4.68	4.06	14 10	1.18 4.59	0.07 5.37	0.16 9.67	0.08 55.29
RV(JB-1	a)	37.60	65.90	7.30	26.00	5.07	1.46	4.67	0.69	3.99	0.71	2.18	0.33	2.10	0.33	3.41	1.93	9.03	1.57	144	26.90	1.44	2.24	0.25
JB-1 RV	Basalt	38.49 <i>38.60</i>	68.07 <i>67.80</i>	7.08 7.01	26.28 26.80	4.97 5.13	1.52 1.49	5.14 <i>4.9</i> 0	0.71 0.82	4.33 4.14	0.81 0.79	2.13 <i>2.27</i>	0.33 0.35	2.14 2.13	0.31 0.31	3.56 3.31	2.51 <i>2.9</i> 3	9.40 <i>9.30</i>	1.67 1.67	136 141	33.98 <i>33.30</i>	1.38 1.33	2.07 1.92	0.27 0.28
JG-2 RV	Granite	19.41 <i>19.9</i> 0	40.14 <i>48.30</i>	6.01 <i>6.20</i>	24.91 <i>26.40</i>	7.67 7.78	0.08 <i>0.10</i>	8.51 <i>8.01</i>	1.53 1.62	10.42 10.50	2.25 1.67	6.13 <i>6.0</i> 4	1.01 1.16	6.90 <i>6.85</i>	1.02 1.22	3.74 4.73	2.02 <i>2.76</i>	29.19 31.60	9.32 11.30	78 96	13.98 14.70	3.25 <i>3.26</i>	2.04 3.00	0.04 <i>0.06</i>
JG-1a	Granite	24.67	52.06	5.75	21.51	4.70	0.72	4.68	0.74	4.76	0.96	2.63	0.43	2.88	0.43	3.92	1.71	14.00	4.75	131	10.85	3.29	3.74	0.06
HV		21.30	45.00	5.63	20.40	4.53	0.70	4.08	0.81	4.44	0.82	2.57	0.38	2.70	U.44	3.59	1.90	12.80	4.69	118	11.40	3.16	4.47	0.05

JB-1b results are six separate digests prepared at the same time as the Tanabe samples. StdDev=Standard deviation (2 sigma); CV%=coefficient of variation. Notes: [1]

No RVs are available for JB-1b, so the comparison is with its predecessor, JB-1a.

[2] [3] JB-1b, JG-2 and JG-1a analysed on an ignited basis, JB-1 hydrous basis.

Recommended values (RV) for GSJ standards from Imai et al. (1995).

# 1. Comparison between XRF and ICP-MS results

Seven elements (Zr, Nb, Th, Ba, Pb, Y and Ce) were determined by both methods. Agreement is excellent for both Zr and Nb (Fig. 3). This is significant, because the possibility of incomplete recovery of HFSE is always present when using dissolution techniques. The comparability

of the Zr and Nb data, and the good agreement between the ICP-MS results for GSJ standards run concurrently and their recommended values (Table 2) suggest that recovery of the HSFE is complete using the digestion methods employed here. Thorium results are also well correlated, but XRF values are uniformly greater (+1.3 ppm) than those

from ICP-MS. This probably reflects a calibration and background bias in the XRF results. The XRF data are also inherently less precise, and the ICP-MS data are thus preferred. To ensure that the Th data in samples not analysed by ICP-MS are comparable, all Th data in Table 1 have been corrected by cross-calibration against the ICP-MS data (Fig. 3).

Ba and Pb analyses by routine solution ICP-MS can be problematic, because of high concentrations even at the dilutions analysed (Ba) or from volatility during acid digestion and alkali fusion (Pb). These features are clearly shown by comparison of the Ba and Pb results (Fig. 4). ICP -MS Ba results above 600 ppm are not plotted, because they are known to be high due to counter saturation. It is evident, however, that this effect also influences results between 500 and 600 ppm, with the ICP-MS results beginning to diverge to higher values. Consequently, the ICP-MS Ba analyses can only be considered accurate below 500 ppm. In contrast, virtually all ICP-MS Pb results are lower than those by XRF (Fig. 4). This strongly suggests loss of Pb during alkali fusion and acid digestion. Because of the above features, the XRF results are preferred, and the ICP-MS Ba and Pb data are not included in Table 2.

Ce and Y results show some differences between the techniques. The relationship for Ce is linear, but the nominal XRF results ex-machine were only comparable with those from ICP-MS between 35 and 65 ppm. Most of the Tanabe samples fall within this range. However, divergence outside of that was significant. ICP-MS results for the GSJ standards compare well with the recommended values (Table 2). This and the complete recovery of HFSE in the ICP-MS AFAD digestion suggest that the problem lay with the XRF data, possibly due to low count rates. Consequently, all Ce data in Table 1 have been crosscalibrated against the Tanabe ICP-MS data to ensure comparability within the dataset. Similar correction to the subset of Shimanto samples analysed at the same time also made that data comparable with the values determined earlier by Roser et al. (1998a).

The ICP-MS yttrium data were consistently low (-20% relative) in the Tanabe samples compared to the XRF results. This was also the case for the GSJ standards (average deviation from recommended values of -16.4%). This is apparently due to methodological bias. Munker (1997; in Robinson et al. 1999) found ICP-MS results to be 10-15% lower than XRF data for the same samples; other workers have also reported low ICP-MS results. Robinson et al. (1999) found a 9% discrepancy between ICP-MS results for USGS standards rocks and published values. However, their XRF data derived from calibrations based on synthetic Y standards agreed with the ICP-MS data.

They interpreted this as reflecting bias towards high values in the recommended results for the USGS standard rocks, which are derived mainly from XRF data acquired from calibrations using natural rock reference materials. This accounts for the methodological discrepancy observed here, and suggests that the recommended values for the GSJ standards are also likely to be systematically high. We have carried out tests using XRF calibrations based on synthetic Y standards, and also find a discrepancy with our XRF calibration built solely on GSJ and USGS standard rocks. This finding agrees with that of Robinson et al. (1999). Our XRF results are thus likely to be systematically high. Because our tests are incomplete and we have not as yet derived precise correction factors, as an interim measure we have adopted the XRF values (Table 1) as analysed, to maintain internal consistency and comparability with the Shimanto data of Roser et al. (1998a) and other published XRF results. The comparability issue will be addressed further in a future paper.

REE results for the GSJ standards compare very well with the preferred values of Imai et al. (1995) (Table 2 and Fig. 5). This verifies that recovery of the HFSE in the alkali fusion and acid digestion was complete.

#### 2. Tanabe Group: General elemental abundances

Traditional variation diagrams using Al<sub>2</sub>O<sub>3</sub> as the abscissa are a useful screening technique to determine the extent of sorting fraction in sedimentary suites, and allow simple comparisons between groups. Example variation diagrams are given in Fig. 6 (major elements) and Figs. 7 and 8 (trace elements), plotted by formation and lithotype (sandstones and mudrocks). These plots also permit comparison with elemental abundances in the Shimanto sediments which are thought to comprise a major part of the Tanabe source.

#### (1) Major elements

The plots show marked sorting fractionation between sand and mud, with a clear division between sandstones and mudrocks, and almost no overlap at around 13.5 wt% Al<sub>2</sub>O<sub>3</sub> (Fig. 6). SiO<sub>2</sub> contents range from about 65% in the most aluminous mudrocks to over 85 wt% in the sandstones. This range is comparable with that seen in the Shimanto terrane Otonashigawa and Muro Groups (Fig. 6), which form the majority of the clasts in Tanabe conglomerates. Most Tanabe samples lie along a linear detrital trend (DT), which is typical of relatively mature quartzose sediments, and are slightly more siliceous at given Al<sub>2</sub>O<sub>3</sub> than their Shimanto protoliths, suggesting increase in detrital quartz during recycling. A few samples, however, scatter to lower values away from DT along dilution lines (DIL), suggesting presence of additional phases which do not contain Si and Al.

TiO<sub>2</sub> forms a tight linear trend, which intersects the



Fig. 3. Comparison of Zr, Nb and Th results for Tanabe Group samples analysed by both XRF and solution ICP-MS. XRF results on the Th plot have been corrected by -1.35 ppm, as outlined in the text.



Fig. 4. Comparison of Ba and Pb results for Tanabe Group samples analysed by both XRF and solution ICP–MS. Samples with Ba >600 ppm are not plotted.

abscissa at around 7% Al<sub>2</sub>O<sub>3</sub> (Fig. 6). This trend differs markedly from that expected from simple silica dilution (SDL), which is sometimes seen in highly mature quartzclay systems or siliceous hemipelagic sediments. The strong correlation between Ti and Al attests to their immobility in geologic systems, and results from sorting fractionation of aluminous clays from sand-sized detritus. Intersection with the abscissa shows that significant amounts of aluminous framework grains (feldspar and rock fragments) remain in the Tanabe sediments. Published modal data for Tanabe sandstones are few, but many modal analyses have been made of coeval and comparable sediments in the nearby Kumano Basin. Kumano Group sandstones average Q68F24R8 (Chijiwa 1992). The major element chemistry of the Tanabe samples is consistent with such values. The Tanabe samples show a similar trend to the Shimanto protolith, but are slightly depleted in TiO<sub>2</sub> at given Al<sub>2</sub>O<sub>3</sub>.



Fig. 5. Comparison of ICP-MS REE results for GSJ rock standards (Table 2) compared to GSJ preferred values. Chondritic normalising values from Taylor and McLennan (1985).

 $K_2O$  (Fig. 6) and MgO also show relatively strong correlation with  $Al_2O_3$ , reflecting primary residence in the clay fraction. For both, however, a small number of samples from the Shirahama Formation scatter to higher values, suggestive of diagenetic or other postdepositional redistribution. The  $K_2O$  plot is also clearly kinked at the sand/ mudrock boundary, similar to the Shimanto protolith. This reflects differing mineralogical control in the sands (K- feldspar) compared to the muds (illitic clays). The Tanabe samples in general are slightly more potassic than the Shimanto suites, tending to plot at the upper end of the distribution. This suggests slightly more intense illitisation.  $Fe_2O_3T$  contents (not illustrated) show weak positive correlation and considerable scatter, but are broadly comparable with the Shimanto protoliths. In detail, Asso Fe data are quite well correlated, lying along what we interpret as the detrital trend. Shirahama data contribute most to the scatter, with contents in the mudrocks being markedly lower than Asso equivalents. This suggests significant provenance change in the Shirahama Formation or postdepositional depletion of Fe.

In contrast, Na<sub>2</sub>O (Fig. 6) and CaO and show weak negative correlation with Al<sub>2</sub>O<sub>3</sub>. This pattern is typical of residence of these elements in plagioclase feldspar, and greater abundances of this mineral in sands than in muds. The Tanabe samples are generally depleted in Na<sub>2</sub>O compared to the Shimanto suites, suggesting loss of albite during source weathering or recycling. A number of samples scatter to higher CaO, reflecting sporadic carbonate enrichment. This accounts for most of the samples which fall on dilution trends on the SiO<sub>2</sub> plot. MnO and P<sub>2</sub>O<sub>5</sub> (not illustrated) display very weak positive correlations, and



Fig. 6. Example major element-Al<sub>2</sub>O<sub>3</sub> variation diagrams (anhydrous normalised basis) for Asso and Shirahama sandstones and mudrocks, compared to probable Shimanto source rocks. Solid line DT= detrital trend (fitted by eye); dashed line SDL= ideal silica dilution trend. Dashed arrows indicate directions of depletion (DIL) or enrichment from the detrital trend. Shaded field: distribution of sandstones and mudrocks from the Nyunokawa, Otonashigawa, and Muro suites of the Shimanto terrane (data of Roser et al. 1998a). Nyunokawa Formation (Hidakagawa Group) samples are included because they are similar chemically to the Otonashigawa Group, and distinct from the more volcanogenic inboard Hidakagawa formations (Yukawa, Terasoma, Miyama and Ryujin).

scatter to depleted values in the mudrocks similar to that shown for Fe. As with Fe, this depletion is a little more marked in the Shirahama Formation.

#### (2) Trace elements

The trace elements analysed by XRF can be divided into three groups on the basis of their trends on variation diagrams. The first and largest group (Ce, Cr, Ga, Nb, Rb, Sc, Ni, Th, Y, and V) shows strong positive correlations with Al, reflecting primary residence in the clay fraction. All of this group except Rb and perhaps Ni, Cr and V are resistant to redistribution by weathering or diagenesis, and hence display the strongest linear correlations. Abundances are also comparable with those in the Shimanto protolith, as shown by plots of Rb, Nb, Th and Y (Fig. 7). The Rb trend is kinked at the sand-mudrock boundary, reflecting contrasting mineralogical control and sorting (K-feldspar in sandstones; illitic clays in mudrocks), and enhanced diagenetic K-metasomatism in the finer sediments. Ga and Nb also exhibit similar but less pronounced kinks. Their immobile nature suggests the cause is related to detrital mineralogy. Ni data in the Shirahama Formation show some depletion in the mudrocks, similar to the pattern shown by Fe and Mn.

The second group (Ba, Pb, Sr) shows considerable scatter, and a tendency for abundances in the sandstones to be equal to or greater those in the mudrocks (Fig. 8). This suggests negative correlation with  $Al_2O_3$ , and hence

association with feldspathic sand-sized detritus. All three elements can also be redistributed during diagenesis, which may have contributed to the scatter. Nevertheless, abundances are broadly similar with the Shimanto protolith suites (Fig. 8).

The third group is represented only by Zr. Asso Formation samples show a linear correlation with Al<sub>2</sub>O<sub>3</sub>, with abundances in mudrocks double those in sandstones (Fig. 8). This pattern is unusual, as Zr abundances in mineralogically mature suites such as these are usually controlled by detrital zircon. This typically produces scattered distributions, with weak maximums in the very fine or fine sands in which zircons are preferentially concentrated. There is no evidence of this in the Asso Formation, implying either residence in the clay fraction, or advanced zircon comminution to very fine size grades coupled with uniform distribution in both sands and mudrocks. Shirahama samples, however, show a more normal pattern, with scatter to high values in sandstones between 10–13 wt% Al<sub>2</sub>O<sub>3</sub>, reflecting zircon concentration. Abundances in Shirahama sandstones are two to three times higher than in Asso equivalents. Mudrocks are also uniformly enriched, but to a lesser extent. Abundances in Shirahama sandstones also closely correspond to those in Shimanto Muro Group sandstones, whereas those in the Asso sands match levels in Otonashigawa and Nyunokawa sandstones (Fig. 8).



Fig. 7. Examples of trace elements, positively correlated with Al<sub>2</sub>O<sub>3</sub> (anhydrous normalised basis). DT and distribution of Shimanto protoliths (shaded field) as in Fig. 7.



Fig. 8. Covariation of Ba, Sr and Zr with Al<sub>2</sub>O<sub>3</sub> (anhydrous normalised basis). Shimanto protoliths (shaded field) on the Ba and Sr plots as in Fig. 7; individual fields for Otonashigawa– Nyunokawa sandstones (shaded) and Muro Group sandstones (dashed field) are differentiated on the Zr plot.

# (3) REE

Chondrite-normalised plots of average REE abundances in average sandstones and mudrocks in each member of the Asso and Shirahama Formations are given in Fig. 9. Average Upper Continental Crust (UCC) and averages for Shimanto sediments are also plotted for comparison. Mudrocks in both the Asso and Shirahama Formations show similar patterns, with relatively high LREE enrichment and moderate europium anomalies. The patterns compare well with UCC and the Shimanto mudrock averages, reflecting derivation from a felsic continental margin source. Significant HREE enrichment (Tm-Lu) is evident in the upper members of the Shirahama Formation however, with patterns climbing above UCC and the Shimanto sandstone averages. This pattern is consistent with concentration of zircons and garnets, which are characteristically enriched in HREE (Taylor and McLennan 1985).

REE in the sandstones are strongly depleted relative to UCC, with La<sub>N</sub> abundances falling to as low as 30 times chondrite, and HREE to as little as 2-3 times. Abundances are similar to or less than those in the average young Shimanto suites plotted. Older Shimanto sandstones (Campanian-Maastricthian and older; Yukawa to Ryujin Formations) have higher REE patterns very similar to UCC. The far lower abundances in the Tanabe averages are suggestive of dilution effect from increase in detrital quartz. This is also reflected by the relative fractionation between sandstone and mudstone, which is more marked in the Shirahama averages than in the Asso data. Almost all the sandstone averages exhibit HREE enrichment, which is particularly pronounced in the Shirahama S 4 and S 5 members. Preliminary REE modeling suggests that the HREE enrichments could be produced by very modest (0.05 and 0.07 volume per cent) concentrations of influxes of zircon and garnet, respectively (Roser et al. 2000). Progressive reduction in Eu anomaly stratigraphically upward is a notable feature of the Shirahama data. At present, the cause for this is unknown. Influx of mafic detritus or concentration of plagioclase and/or epidote could produce this effect, but there is no petrographic evidence to support these options.

#### **Elemental Abundances: Significance**

General elemental abundances and trends show marked sorting fractionation, as expected from the mineralogical and sedimentological maturity of the Tanabe sediments. The trends produced are consistent with separation of quartzfeldspar-lithic sand-sized detritus from aluminous silt and mud. The trends, extent of fractionation, and the elemental groupings shown by the variation diagrams are comparable with the Shimanto Otonashigawa and Muro Group sediments which were large components of the Tanabe source. In detail, however, some contrasts are evident. Fractionation is more advanced in the Shirahama Formation than in the Asso Formation, as shown by greater spread in abundances. For most elements, nearly all Shirahama mudrocks have higher abundances than Asso equivalents. Shirahama sandstones also spread to lower abundances than the Asso sands. This increased fractionation between sand and mud is also clearly shown by the REE, and is likely to be a product of winnowing in the storm-dominated shelf environment of the upper part of the Tanabe Group. Contrasting behaviour of Zr also show that significant differences occur in chemistry between the transgressive and regressive parts of the Tanabe sequence. These features suggest that significant modification of the bulk chemistry of clastic sediments can occur in shallow water forearc



Fig. 9. Chondrite-normalised plots of average REE abundances in sandstones and mudrocks from the Tanabe Group, compared to UCC and Shimanto averages. Maro and Hisogawa averages include data for time-equivalent Gohchidani samples. Chondritic values and UCC (average Upper Continental Crust) from Taylor and McLennan (1985). Shimanto averages from Roser & Kimura (unpublished data): Grp 2=Nyunokawa Formation and Otonashigawa Group; Grp 3= Muro Group.

settings. The nature of these modifications and their implications will be discussed in more detail in a future paper.

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# (要旨)

Barry Roser・木村純一・久富邦彦,2001,紀伊半島田辺層群の砂岩と泥質岩の全岩元素組成,島根 大学地球資源環境学研究報告,19,101-112

珪質砕屑岩類である下部中新統の田辺層群は、和歌山県紀伊半島の南西部に露出する.この田辺 層群は、上昇した四万十付加帯コンプレックスの上位に堆積した前弧盆のサクセッションである. 本論では、田辺層群の層序に沿って採取した砂岩および泥質岩 87 試料の全岩蛍光 X 線分析結果を 報告する.さらに、54 試料を酸分解-アルカリ融解 ICPMS 法によって分析した.田辺層群の基部 をなす朝来累層の分析結果は砂岩と泥質岩の間で特徴ある元素分析を示し、その傾向は上位の白浜 累層でより顕著になる.これはおそらく暴浪の卓越した沿海環境での洗い出しによるとみられる. また、元素濃度は全般的に田辺層群の供給源となっている四万十帯のプロトリスと良く似ている.